## A Novel Synthesis of 1,3-Diol Diesters by the Reaction of Aldehydes with Oxime Esters Catalyzed by Samarium Complex

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 $Cp*_2Sm(thf)_2$  was found to be an efficient catalyst for the synthesis of 1,3-diol diesters by the coupling reaction of aldehydes with oxime esters under mild conditions. For instance, the reaction of acetaldehyde with cyclohexanone oxime acetate catalyzed by  $Cp*_2Sm(thf)_2$  gave 1,3-diacetoxybutane in 70% yield. Treatment of acetaldehyde with isopropenyl acetate in the presence of a small amount of cyclohexanone oxime acetate and  $Cp*_2Sm(thf)_2$  resulted in the corresponding 1,3-diol diester in good yield.

Samarium compounds have been widely used in organic synthesis. Recently, it has been reported that  $SmI_2$  catalyzes intramolecular Tishchenko reaction,  $^1$  epoxide rearrangements,  $^2$  Michael and aldol reactions,  $^3$  and Diels-Alder reaction.  $^4$  Samarium complex such as  $Cp\ast_2Sm(thf)_2$  is also a useful catalyst for hydrogenation  $^5$  and hydroboration of alkenes  $^{6,7}$  and hydroamination/cyclization of amines.  $^{5g,6}$ 

Previously, we reported that Cp\*2Sm(thf)2 and SmI2 were efficient catalysts for a new type of 1:2 cross coupling reaction of aldehydes with vinyl esters under mild conditions to form the corresponding 1,3-diol diesters in moderate to good yields (eq. 1).8

Now we find that Cp\*2Sm(thf)2 catalyzes the coupling reaction of oxime esters with aldehydes to form 1,3-diol diesters.

The reaction of cyclohexanone oxime acetate (1) with butyraldehyde (2a) was examined in the presence of some samarium compounds (eq. 2 and Table 1).

A typical reaction was carried out as follows: To a mixture of aldehyde (3 mmol) and oxime acetate (1 mmol) was added  $Cp*_2Sm(thf)_2$  (0.1 mmol) in toluene solution (1 mL). After stirring the solution at 50 °C for 3 h, the reaction mixture was quenched with wet ether. Products were purified by column chromatography on silica gel with n-hexane / ethyl acetate (10 / 1 v/v).

The reaction of **1** with 3 equiv of **2a** catalyzed by  $Cp*_2Sm(thf)_2$  (10 mol% with respect to **1**) at 50 °C for 3 h gave 3-acetoxymethyl-4-butanoylheptane (**3a**) in 68% yield along with cyclohexanone oxime (**4**) (60%) (run 1). The reaction proceeded smoothly even in the presence of 5 mol% of  $Cp*_2Sm(thf)_2$  (run 2). **3a** was obtained in 77% yield when **1** was reacted with 2 mmol of **2a** (run 3).

**Table 1.** Reaction of cyclohexanone oxime acetate (1) with butyraldehyde (2a) catalyzed by Sm compounds<sup>a</sup>

Run	Catalyst	Solvent	<b>3a</b> (Yield /%)
1	$Cp*_2Sm(thf)_2$	toluene	68
2 <sup>b</sup>	$Cp*_2Sm(thf)_2$	toluene	68
3 <sup>c</sup>	$Cp*_2Sm(thf)_2$	toluene	77 <sup>d</sup>
4	$Cp*_{2}Yb(thf)_{2}$	toluene	62
5	$Sm(O^iPr)_3$	THF	56
6	$SmI_2$	THF	<del>-</del>
7	$SmI_3$	THF	
8	Sm(OTf) <sub>3</sub>	THF	

<sup>a</sup>**1** (1 mmol) was reacted with **2a** (3 mmol) in the presence of catalyst (0.1 mmol) in solvent (1 mL) at 50 °C for 3 h. <sup>b</sup>Cp\*<sub>2</sub>Sm(thf)<sub>2</sub> (0.05 mmol) was used. <sup>c</sup>**2a** (2 mmol) was used. <sup>d</sup>Based on **2a**.

Sm(O<sup>i</sup>Pr)<sub>3</sub> and Cp\*<sub>2</sub>Yb(thf)<sub>2</sub> also catalyzed the present reaction, but their catalytic activities were slightly lower than that of Cp\*<sub>2</sub>Sm(thf)<sub>2</sub> (runs 4 and 5). SmI<sub>2</sub>, SmI<sub>3</sub> and Sm(OTf)<sub>3</sub> were inert for this reaction (runs 6-8).

On the basis of these results, a variety of aldehydes (**2b-2f**) were allowed to react with **1** in the presence of Cp\*<sub>2</sub>Sm(thf)<sub>2</sub> catalyst in toluene at 50 °C for 3 h. Representative results are shown in Table 2.

Linear aldehydes **2b** and **2c** were reacted with **1** to give the corresponding 1,3-diesters **3b** and **3c** in 70% yields, respectively

**Table 2.** Reaction of cyclohexanone oxime acetate (1) with various aldehydes catalyzed by Cp\*<sub>2</sub>Sm(thf)<sub>2</sub> <sup>a</sup>

Run	Aldehyde	Product (Yield /%)	
	R^CHO	O R O R	
$1^{b}$	<b>2b</b> (R=H)	Ŕ <b>3b</b> (70)	
2	<b>2c</b> (R=Me)	<b>3c</b> (70)	
3	<b>2d</b> (R= <sup>i</sup> Pr)	<b>3d</b> (54)	
4	СНО	$ \begin{array}{cccc}  & & & & & & & & \\  & & & & & & & \\  & & & &$	
	2e	, З <b>е</b>	
5 <sup>b</sup>	CHO 2f	O Hex O CHex (44)	

<sup>&</sup>lt;sup>a</sup>**1** (1 mmol) was reacted with aldehyde (3 mmol) in the presence of Cp\*  ${}_2$ Sm(thf) ${}_2$  (0.1 mmol) in toluene (1 mL) at 50 °C for 3 hr.

<sup>&</sup>lt;sup>b</sup>The reaction was carried out at room temperature.

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(runs 1 and 2). Although branched aldehyde such as 3-methylbutyraldehyde (**2d**) and isobutyraldehyde (**2e**) reacted similarly with **1** to give diester, **3d** and **3e**, in 54% and 76% yields (runs 3 and 4). However, the reaction of cyclohexanecarboxaldehyde (**2f**) with **1** resulted in **3f** in lower yield because of the concomitant formation of Tishchenko product, (cyclohexyl)methyl cyclohexanecarboxylate (21%) (run 5).

To clarify the reaction path for the present 1:3 coupling reaction of oxime ester with aldehyde, the deuterium label experiment was performed. The reaction of **1** with 3 equiv of  $CD_3CDO$  (**2b**-d) catalyzed by  $Cp*_2Sm(thf)_2$  produced **3b**-d in which 11 deuteriums are incorporated into the molecule (eq. 3). <sup>9</sup>

This fact shows that 1 reacts with  $2\mathbf{b}$ -d to form vinyl acetate [A] as a transient intermediate, which then couples with  $2\mathbf{b}$ -d to form 1,3-diol diester  $3\mathbf{b}$ -d (Scheme 1).

CD<sub>3</sub>CDO

$$\begin{array}{c}
cat. \text{ Cp*}_2\text{Sm(thf)}_2\\
\hline
1 & DON \\
\hline
\end{array}$$

$$\begin{array}{c}
cat. \text{ Cp*}_2\text{Sm(thf)}_2\\
\hline
\end{array}$$

$$\begin{array}{c}
cat. \text{ Cp*}_2\text{Sm(thf)}_2\\
\hline
\end{array}$$

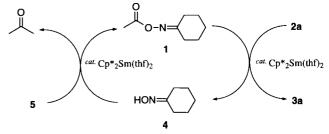
$$\begin{array}{c}
3\mathbf{b}\text{-}d\\
\end{array}$$

**Scheme 1.** A possible reaction path for the reaction of **2b**-d with **1** catalyzed by  $Cp*_2Sm(thf)_2$ 

It is interesting to note that the Cp\*<sub>2</sub>Sm(thf)<sub>2</sub>-catalyzed 1:3 coupling reaction of isopropenyl acetate (5) with aldehyde **2a** was enhanced by adding a small amount of oxime ester **1**. The reaction of **5** (1 mmol) with **2a** (3 mmol) in the presence of **1** (0.2 mmol) by Cp\*<sub>2</sub>Sm(thf)<sub>2</sub> (0.1 mmol) in toluene (1 mL) at 50 °C for 6 h formed **3a** in 61% yield, while the reaction in the absence of **1** under these conditions gave **3a** in only 17% yield (eq. 4).

The oxime ester-mediated coupling reaction of **5** with **2a** is thought to proceed as follows (Scheme 2).

Diester  $\bf 3a$  seems to be formed by the reaction of  $\bf 2a$  with  $\bf 1$  rather than with  $\bf 5$ , and the resulting oxime  $\bf 4$  is considered to be smoothly acylated with isopropenyl acetate  $\bf 5$  in the presence of  $\rm Cp^*_2Sm(thf)_2$  to regenerate  $\bf 1$ . Indeed,  $\bf 4$  readily acylated by  $\bf 5$  under the influence of  $\rm Cp^*_2Sm(thf)_2$  at room temperature giving  $\bf 1$  in



Scheme 2. Oxime ester-mediated coupling reaction of isopropenyl acetate (5) with aldehyde by Cp\*<sub>2</sub>Sm(thf)<sub>2</sub>

quantitative yield.

In summary, a novel synthesis of 1,3-diol diesters by the coupling reaction of oxime ester with aldehydes was achieved in the presence of a catalytic amount of  $Cp*_2Sm(thf)_2$ . The coupling reaction of isopropenyl acetate **5** with aldehyde was found to be facilitated by adding a small amount of oxime ester **1** 

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- 9 Spectral data for **3b**-d is as follows: <sup>1</sup>H-NMR (CDCl<sub>3</sub> / Me<sub>4</sub>Si)  $\delta$  1.98 (s); <sup>13</sup>C-NMR (CDCl<sub>3</sub>)  $\delta$  18.4, 18.7,18.9, 19.2, 19.5, 19.9, 20.1, 20.4, 20.8, 21.0, 21.1, 33.0, 33.2, 33.5, 33.8, 34.1, 59.3, 59.6, 60.0, 60.2, 60.6, 66.8, 67.1, 67.4, 170.5, 170.9.